

ASYMMETRIC SYNTHESIS OF CIS-2-SUBSTITUTED
 CYCLOHEXANAMINES WITH HIGH OPTICAL PURITY

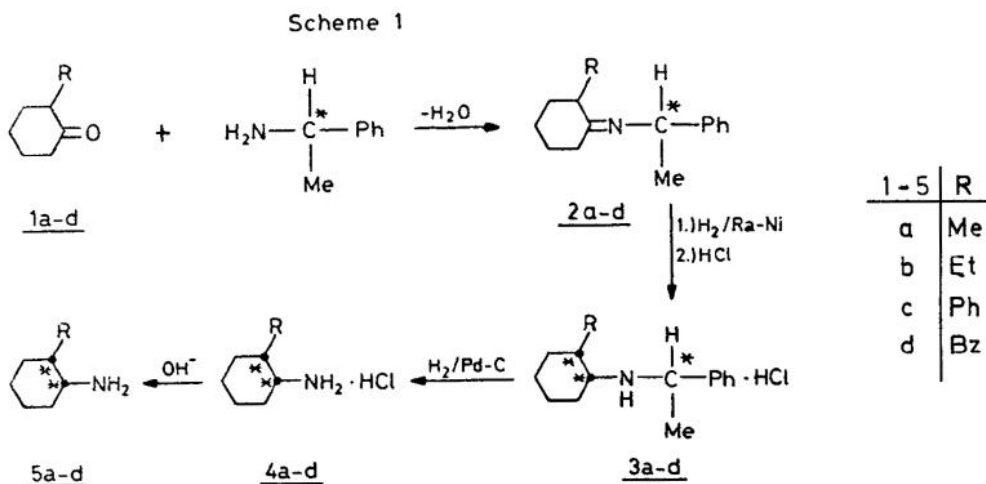
A.W. Frahm[†] and G. Knupp

Institut für Pharmazeutische Chemie der Universität
 Kreuzbergweg 26, 5300 Bonn 1, W-Germany

Abstract: Asymmetric reductive amination of racemic 2-substituted cyclohexanones (R= methyl, ethyl, phenyl, benzyl) using optically active 1-phenylethylamines yields optically active cis-cyclohexanamines.

The nonenzymatic asymmetric synthesis of open-chain amines and amino acids from the corresponding ketones or keto acids is well documented¹⁻⁵. Hiskey⁶ and later on Harada⁷ were the first who synthesized optically active amino acids from keto acids by hydrogenolytic asymmetric transamination in good optical yield. Barfknecht⁸ and Standridge⁹ showed the general applicability of this reaction in the synthesis of psychotomimetic phenyl-isopropylamines and analogues of mescaline (optical purities between 95 and 100%).

We report now the high-yield asymmetric synthesis of optically active cis-2-substituted cyclohexanamines 5a-d from racemic 2-substituted ketones 1a-d (Scheme 1), which is, by our knowledge, the first asymmetric synthesis of carbocyclic amines from corresponding cyclanones.



It is remarkable, that the hydrogenation leads to optically pure *cis*-products (enantiomeric excess >92%), thus running under both highly diastereoselective and enantioselective control.

Reaction of racemic cyclohexanones 1a-d with either R-(+)- or S-(-)-1-phenylethylamine in a Dean-Stark-apparatus yields azomethines 2a-d, which are immediately hydrogenated over Raney-nickel in a Parr-shaker at 5 bar. The resulting secondary amines are isolated as hydrochlorides 3a-d in good chemical yield (Table 1).

TABLE 1: PROPERTIES OF SECONDARY AMINE-HYDROCHLORIDES 3a-d ^a

Com- pound	Starting amine	Yield ^b [%]	m. p. [°C]	$[\alpha]_D^{23}$ (c=1.2/EtOH)
<u>3a</u>	+	88	279-80	+ 52.8
	-		280-81	- 52.9
<u>3b</u>	+	90	245-47	+ 52.9
	-		245-47	- 53.3
<u>3c</u>	+	49 ^c	--- ^d	---
	-		--- ^d	---
<u>3d</u>	+	80	256-59	+ 85.7
	-		257-59	- 83.7

^a spectroscopical data and microanalyses are in agreement with structures proposed; ^b based on starting ketones; ^c yield of N-(2-phenylcyclohexan)-1-phenylethylamine-hydrochlorid 3c is significantly lower than that of the other amines because of partly isomerisation of azomethine 2c to the corresponding enamine, which is not hydrogenated under the reaction conditions; ^d liquid.

Without further purification 3a-d are hydrogenolized with palladium-on-charcoal catalyst (5%) at 5 bar and 45°C in a Parr-shaker to yield optically active 2-substituted cyclohexanamine-hydrochlorides 4a-d in good chemical and optical yield (Table 2). Hydrochlorides are liberated to free bases 5a-d by aqueous sodium hydroxide quantitatively for spectroscopical purposes only (fast carbonisation occurs!).

TABLE 2: PROPERTIES OF PRIMARY AMINE-HYDROCHLORIDES 4a-d^a

Com- pound	Starting amine	Yield ^b [%]	m. p. [°C]	$[\alpha]_D^{23}$ (c=1.3/EtOH)	enantiomeric excess [%]
<u>4a</u>	+	78	234-35	+ 8.0	96
	-		234-35	- 8.1	97
<u>4b</u>	+	79	193-95	- 0.5	92
	-		193-94	+ 0.6	94
<u>4c</u>	+	42	248	- 106.6	94
	-		247-48	+ 104.4	92
<u>4d</u>	+	68	272-74	+ 14.3	99
	-		272-74	- 14.0	98

^a spectroscopical data and microanalyses are in agreement with structures proposed and literature data²¹⁻²³; ^b overall yield

The purity of the secondary and primary amines was checked by chromatographical and spectroscopical methods. Neither TLC and HPLC nor ¹H- and ¹³C-NMR showed any traces of the trans-amines. The optical purity of the corresponding enantiomers was checked by HPLC. For this purpose 4a-d were acylated by the method of Dale¹⁰ with optical pure (+)-2-methoxy-2-trifluoromethyl-phenylacetic acid chloride (MTPA-Cl) to the diastereomeric amides. In all cases the enantiomeric excess was higher than 92%.

Under above conditions only the thermodynamically less stable cis-compounds are isolated. This is consistent with Barton's rule¹¹, although it is known, that under more vigorous hydrogenation conditions (i.e. 80 bar, 100°C) always the thermodynamically more stable products dominate in comparable cases^{12,13}.

With exception of compound 4c, cis configurated compounds 3a-d and 4a-d are synthesized for the first time in its enantiomeric forms¹⁴. Although racemic and optically active trans-2-methyl-cyclohexanamines are well known¹⁵⁻¹⁷, from the corresponding cis-compounds only the racemic modification is documented¹⁸⁻²⁰.

The absolute configuration of the synthesized compounds is not yet established. CD- and X-ray-analyses are in progress and will be published later.

In forth-coming publications we shall show the result of our investigations, which deal with cyclanones of different ring sizes and R-groups in changing positions.

We thank the Fonds der Chemischen Industrie and the BASF-Ludwigshafen for supporting this work.

REFERENCES

1. F. Knoop and C. Martius, *Zeitschr. Physiol. Chem.* 258, 238 (1939).
2. J.B. Longenecker and E.E. Snell, *Proc. Nat. U.S. Acad. Sci.* 42, 221 (1956).
3. S. Akabori, S. Sakurai, Y. Izumi and Y. Fujii, *Nature* 178, 323 (1956).
4. O. Cervinka, V. Suchan, O. Kotynek and V. Dudek, *Collect. Czech. Chem. Commun.* 30, 2484 (1965).
5. F. Weinges and G. Graab, *Chem. Ztg. Chem. App.* 94, 728 (1970).
6. R.G. Hiskey and R.C. Northrop, *J. Am. Chem. Soc.* 84, 4798 (1962).
7. K. Harada, *Nature* 212, 1571 (1966).
8. D.E. Nichols, C.F. Barfknecht, D.B. Rusterholz, F. Benington and R.D. Morin, *J. Med. Chem.* 19, 480 (1973).
9. R.T. Standridge, H.G. Howell, J.A. Gylys, R.A. Partyka and A.P. Shulgin, *J. Med. Chem.* 19, 1400 (1976).
10. J.A. Dale, D.L. Dull and H.S. Mosher, *J. Org. Chem.* 34, 2543 (1969).
11. D.H.R. Barton, *J. Chem. Soc.* 1954, 1027.
12. M. Murakami, K. Suzuki, M. Fujishige and J.-W. Kang, *Nippon Kagaku Zasshi*, 85, 235 (1964); *C.A.* 61 (1965), 13408.
13. L.K. Freidlin, E.F. Litvin, V.V. Yakubenok and L.P. Pivonenkova, *Izv. Akad. Nauk SSSR, Ser. Khim.* 1973, 850; *C.A.* 79 (1973), 52904.
14. L. Verbit and H.C. Price, *J. Am. Chem. Soc.* 94, 5143 (1972).
15. M. Mousseron and P. Froger, *Bull. Soc. Chem. Fr.* 14, 843 (1947).
16. H. Nohira, K. Ehara and A. Miyashita, *Bull. Chem. Soc. Jpn.* 43, 2230 (1970).
17. W. Hüchel and K.D. Thomas, *Justus Liebigs Ann. Chem.* 645, 177 (1961).
18. Z.J. Vejdelek, M. Rajsner and M. Protiva, *Collect. Czech. Chem. Commun.* 25, 245 (1960); for comparison we repeated this investigation and have found, that the postulated cis-2-methyl-cyclohexanamine is a mixture of trans and cis product in the ratio of 64:36 .
19. H. Feltkamp, *Arch. Pharm.* 295, 764 (1962).
20. H. Booth, G.C. Gidley and N.C. Franklin, *Tetrahedron* 23, 2421 (1967).
21. H. Feltkamp and K.D. Thomas, *Justus Liebigs Ann. Chem.* 683, 49 (1965).
22. W. Naegele and D. Wendisch, *Org. Magn. Reson.* 2, 439 (1970).
23. D. Dodrell, I. Burfitt and N.V. Riggs, *Aust. J. Chem.* 28, 369 (1975).

(Received in Germany 6 April 1981)